

Title: Alkali metal also known as

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Li, Na, K, Rb, and Cs are all group IA elements, also known as the alkali metals. The seventh member of the group, francium (Fr) is radioactive and ...

Alkali metals (also known as the lithium family) are a group of six elements on the periodic table that include lithium (Li), sodium (Na), potassium ...

The alkali metals, also known as the alkali metal family is a group of six elements characterized by common physical and chemical properties, a similar electron ...

Alkali metals are the six different chemical elements found in the first column of the periodic table: lithium (Li), sodium (Na), potassium (K), rubidium ...

Alkali metals are a group of six highly reactive metallic elements found in Group I of the periodic table, comprising lithium (Li), sodium (Na), potassium (K), rubidium (Rb), cesium (Cs), and francium (Fr).

Alkali metals are the six elements that comprise Group I in the Periodic Table: lithium (Li), sodium (Na), potassium (K), rubidium (Rb), cesium (Cs), and francium (Fr).

When they react with water, Group 1 elements form metal hydroxides which are alkaline (with a pH above 7). This is why they are called alkali metals.

In simple words, those metals which form a strong base (alkalis or alkaline solution) on reacting with water, are known as alkali metals. Let's ...

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